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Sir Periodic Table

of Elements

Explained - Metals,

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Nonmetals,  
Valence Electrons,  
Charges Who

REALLY invented  
the periodic table?

Interpreting The  
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Answers

Read the periodic  
table from top left  
to bottom right.

The elements are  
ordered by their  
atomic numbers,

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which increase as you move across and down the periodic table. The atomic number is how many protons the element's atom possesses. You'll also notice that each element's atomic mass increases as you move across the table.

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in each square of  
the periodic table:  
Its atomic number,  
chemical symbol,  
name, and atomic

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Answers  
mass. Decide if  
each of the  
underlined terms is  
used correctly. If it  
is, write correct  
after the sentence.

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The Periodic Table  
can provide a  
wealth of

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Answers

information if it is interpreted properly. Click on the video below to understand and interpret the arrangement of elements in the Periodic Table. Be sure to click on the "full screen" icon. Record the information in your notebook or on the

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The Arrangement  
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Q5 Based on a  
typical modern  
version of the  
Periodic Table (a)  
In order of atomic  
(proton) number.  
(b) Due to isotopic

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masses, the  
relative atomic  
mass does go  
'down' occasionally  
(there is no  
obvious 'nuclear'  
rule that applies).  
BUT chemically Te  
is S and Se etc. and  
I is like Cl and Br  
etc.

gcse ANSWERS to  
the Periodic Table



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The periodic table is useful for understanding atomic properties that show periodic trends. One such property is the atomic radius (Figure 2.7. 5). The atomic radius is defined as one-half the distance

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between the nuclei of identical atoms that are bonded together. The units for atomic radii are picometers, equal to  $10^{-12}$  meters.

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